

# High-pressure Metal Hydride Tank for Fuel Cell Vehicles

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- Hydrogen storage capacity

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# Issues on FCV towards Market Introduction

## Major Issues

Items	Challenges	Responsibility
<b>Technical</b>	<b>Low temperature, High temperature, High Efficiency · Size reduction, Reliability · Durability, Salt water, Dust, Volcanic gas (H<sub>2</sub>S), High-altitude, electro-magnetic wave, etc.</b>	<p><b>Vehicle Manufacturers</b></p> <p><b>Government, Energy Supplier</b></p>
<b>Marketability</b>	<b>Driving range (Hydrogen storage), Cost (Vehicle cost)</b>	
<b>Environment</b>	<b>Recyclability, Life Cycle Assessment (LCA)</b>	
<b>Safety</b>	<b>Hydrogen, High voltage, Crash worthiness</b>	
<b>Infrastructure</b>	<b>Hydrogen production · transportation · storage, Infrastructure development, Hydrogen cost</b>	



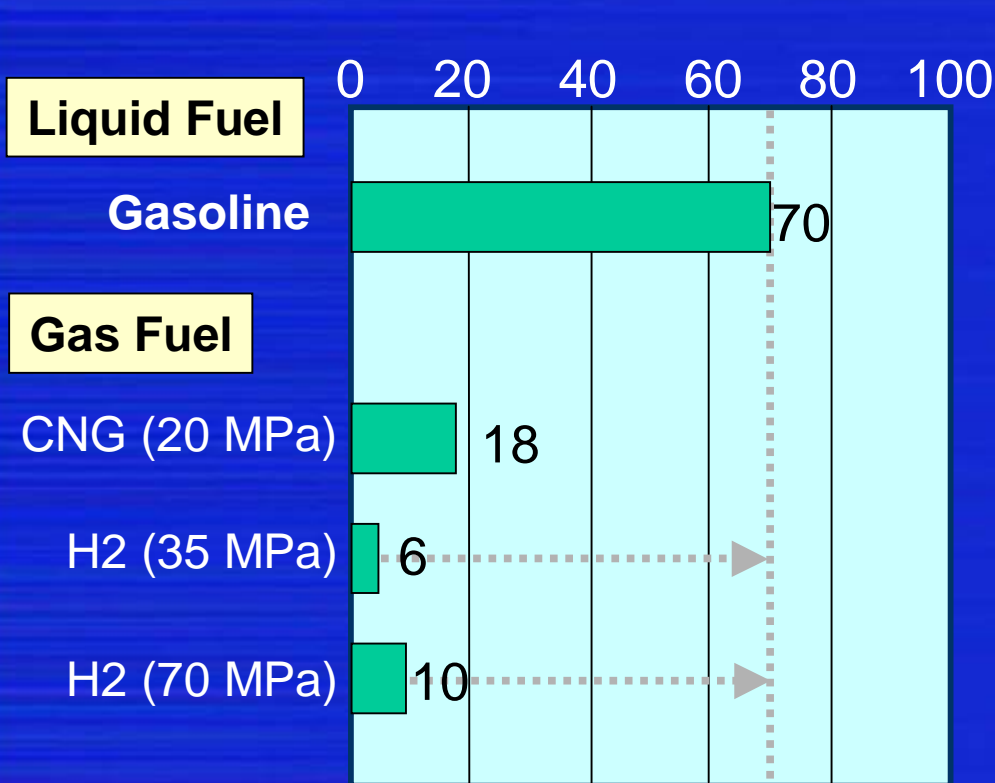
# Cruising Range of High-pressure Storage

High-pressure storage is not sufficient to provide enough energy density

## (1) Comparison of fuel amounts

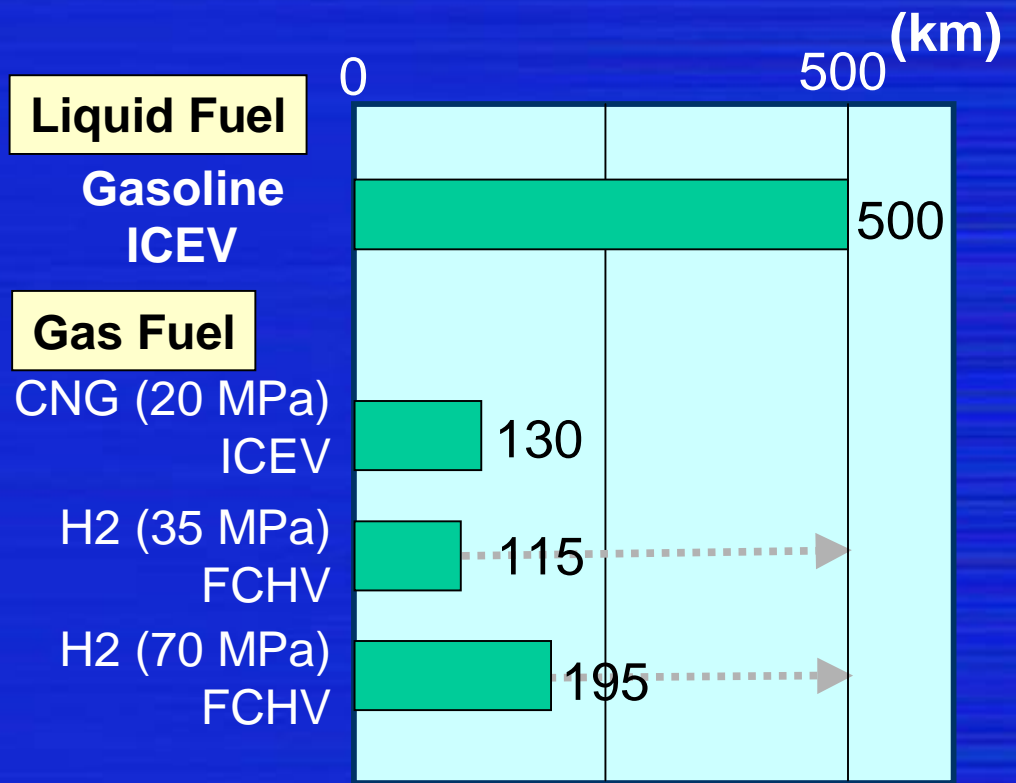
(Tank capacity of 70L)

Gasoline equivalent (L)

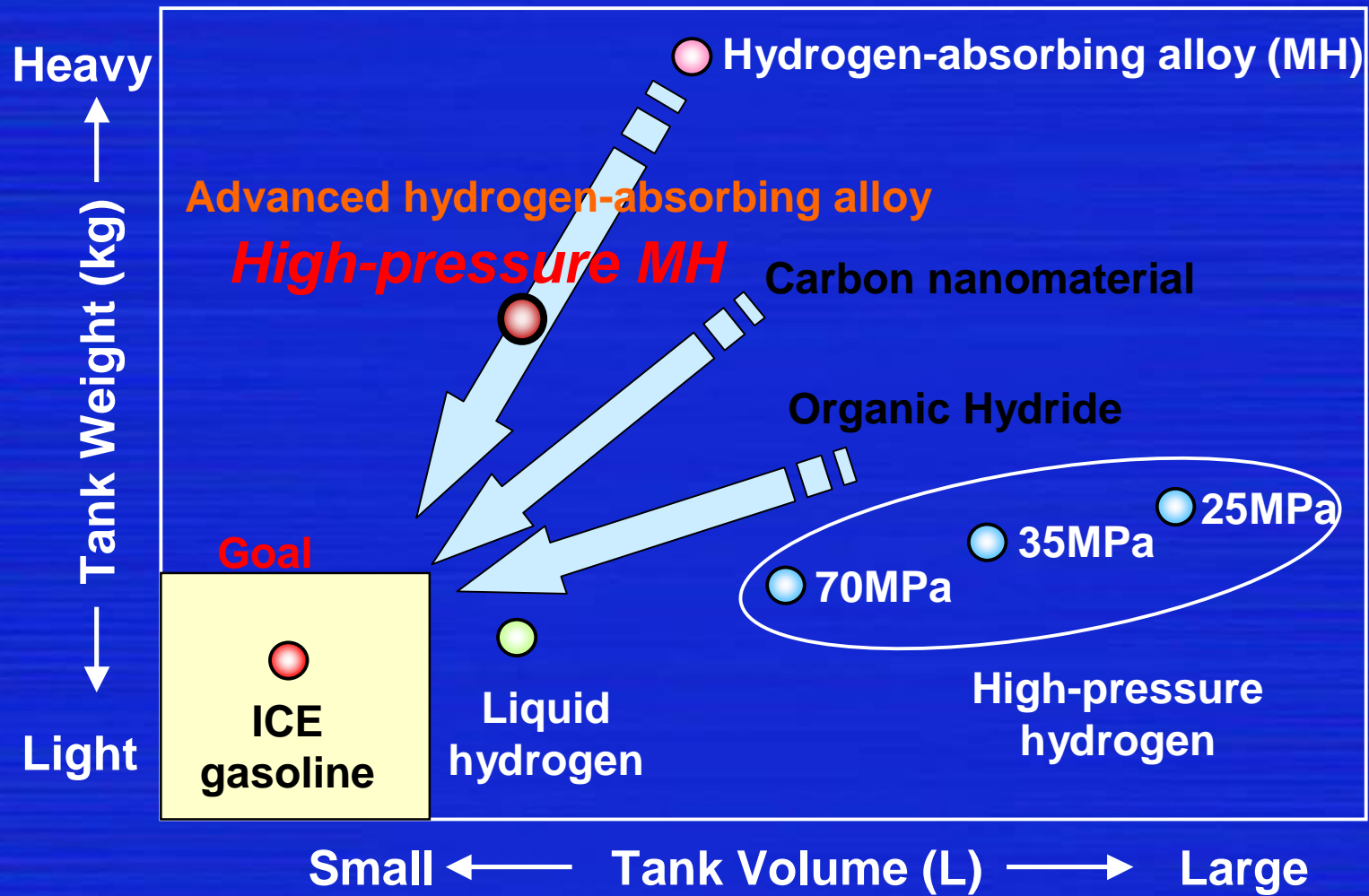


## (2) Comparison of ranges

(Tank capacity of 70L)



# Hydrogen Storage Technology

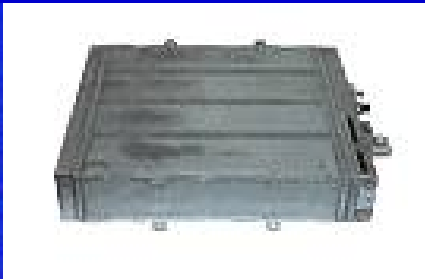


# Hydrogen Tank for FCHV

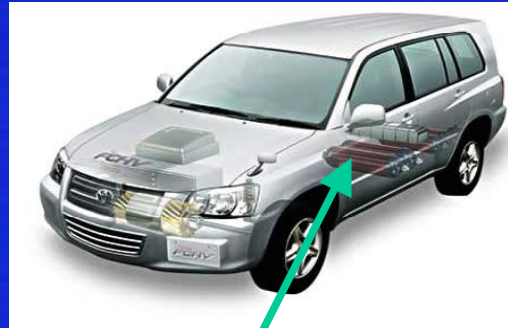
FCHV-3 (2001)



**Metal Hydride Tank**  
(Low-pressure system with Ti-Cr-V alloy)



Toyota FCHV (2002)



**35 MPa High-pressure Hydrogen Tank**



**70 MPa High-pressure Hydrogen Tank\***  
(Developed in Toyota)



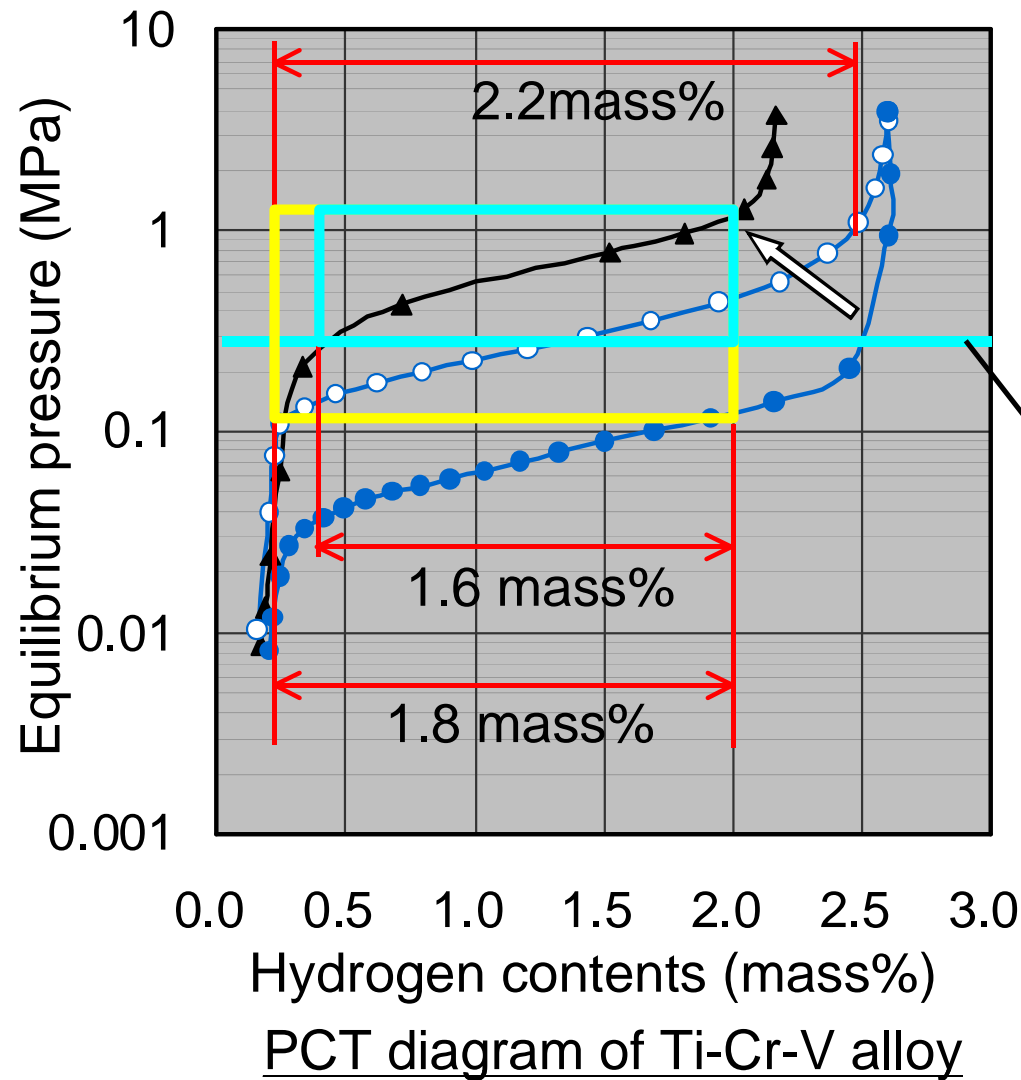
**35 MPa High-pressure Hydrogen Tank\***  
(Developed in Toyota)



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\*Ref. M. Mizuno, et al., Toyota Motor Corp., Proceedings of the 2005 Spring Meeting of JSAE, EV·HEV·FCV Systems-Components/Evaluation

# Issues of Low-pressure MH system



Available hydrogen storage capacity decreased by various restrictions.

- Restriction of temperature-pressure band
- Absorption → desorption hysteresis
- To keep system performance
- Low-temperature

Required pressure to supply H<sub>2</sub> for FC system

desorption (308 K)

absorption (268 K)

desorption (268 K)



# Performance of On-board Tank System

	Low-pressure MH tank Ti-Cr-V System	High-pressure tank
Hydrogen storage capacity	3.5 kg /tank 120 L	3 kg / tank 180 L
Tank weight	300 kg	< 100 kg
Hydrogen filling time	30 -60 min. With external cooling facility	5-10 min.
Hydrogen release at low temperature	Difficult under 308 K	Possible
Control ability	Difficult in acceleration	Good
Safety	Low pressure (<1 MPa)	High-pressure (35 MPa)

According to our experience...

**External cooling during refueling**

Is not easy  
for example liquid connection

**For on-board heating during release**

Only generated heat in FC stack  
is available



# High-pressure MH Tank

- High pressure cylinder vessel with MH and built in heat exchanger

## *Metal hydride*

Ti-Cr-Mn\* (AB<sub>2</sub> laves phase)  
Hydrogen amount: 1.9 mass%  
| H<sup>0</sup> |: 22kJ/molH<sub>2</sub>  
Desorbing pressure:  
0.5MPa at 243K

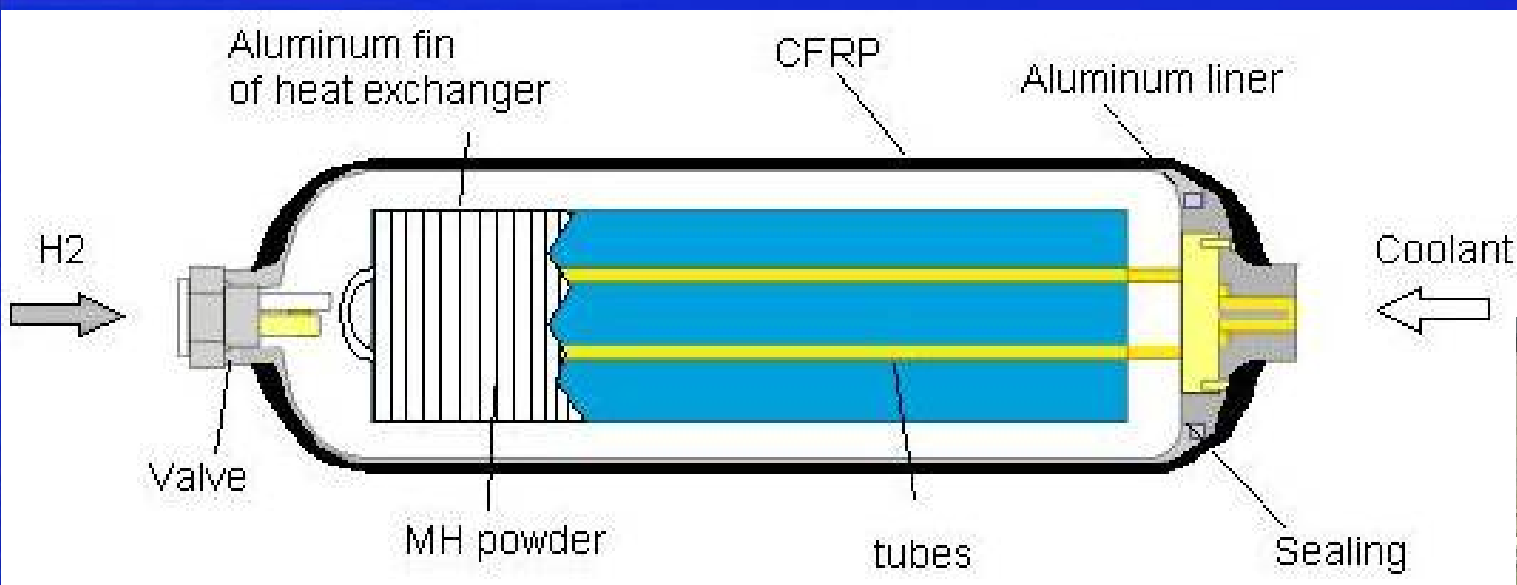
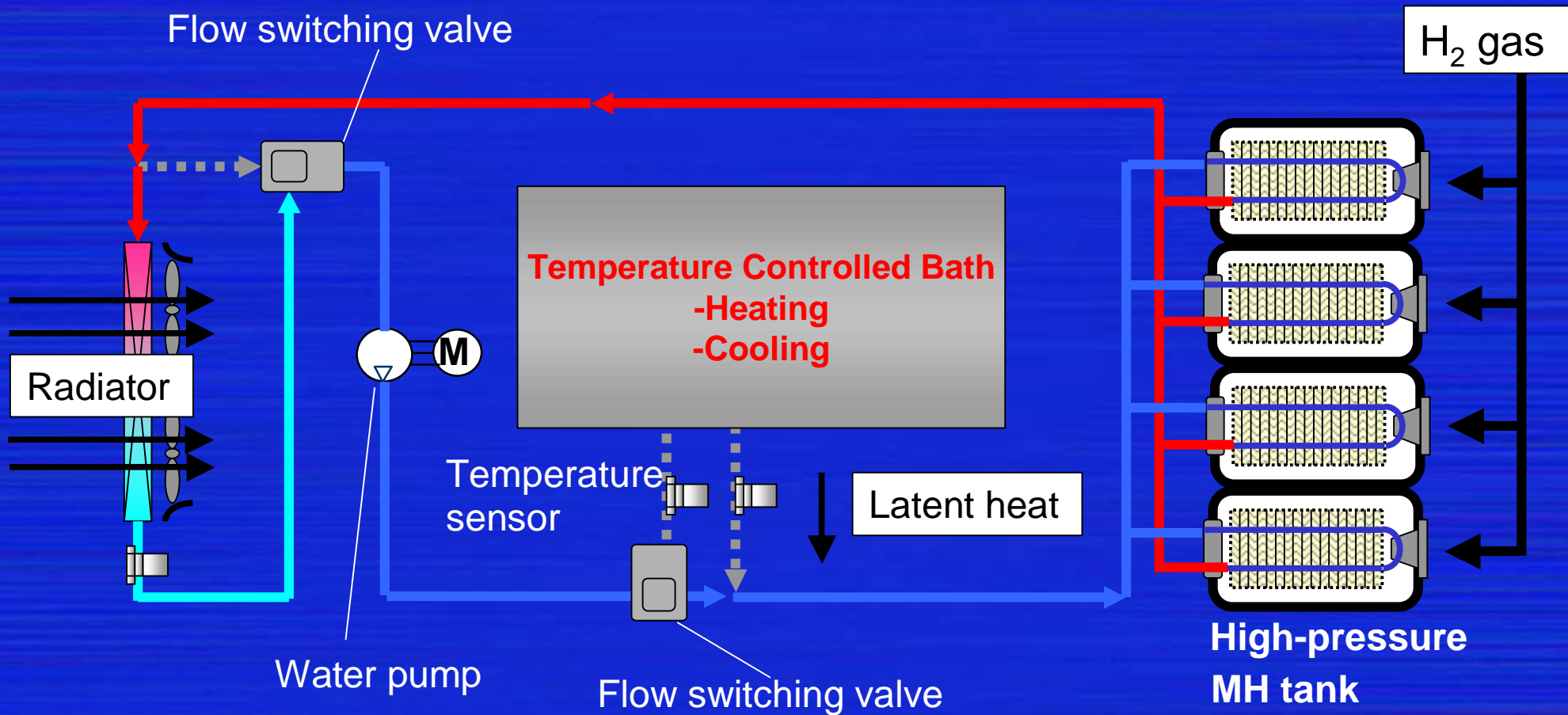


Fig. Schematic view of high-pressure MH tank

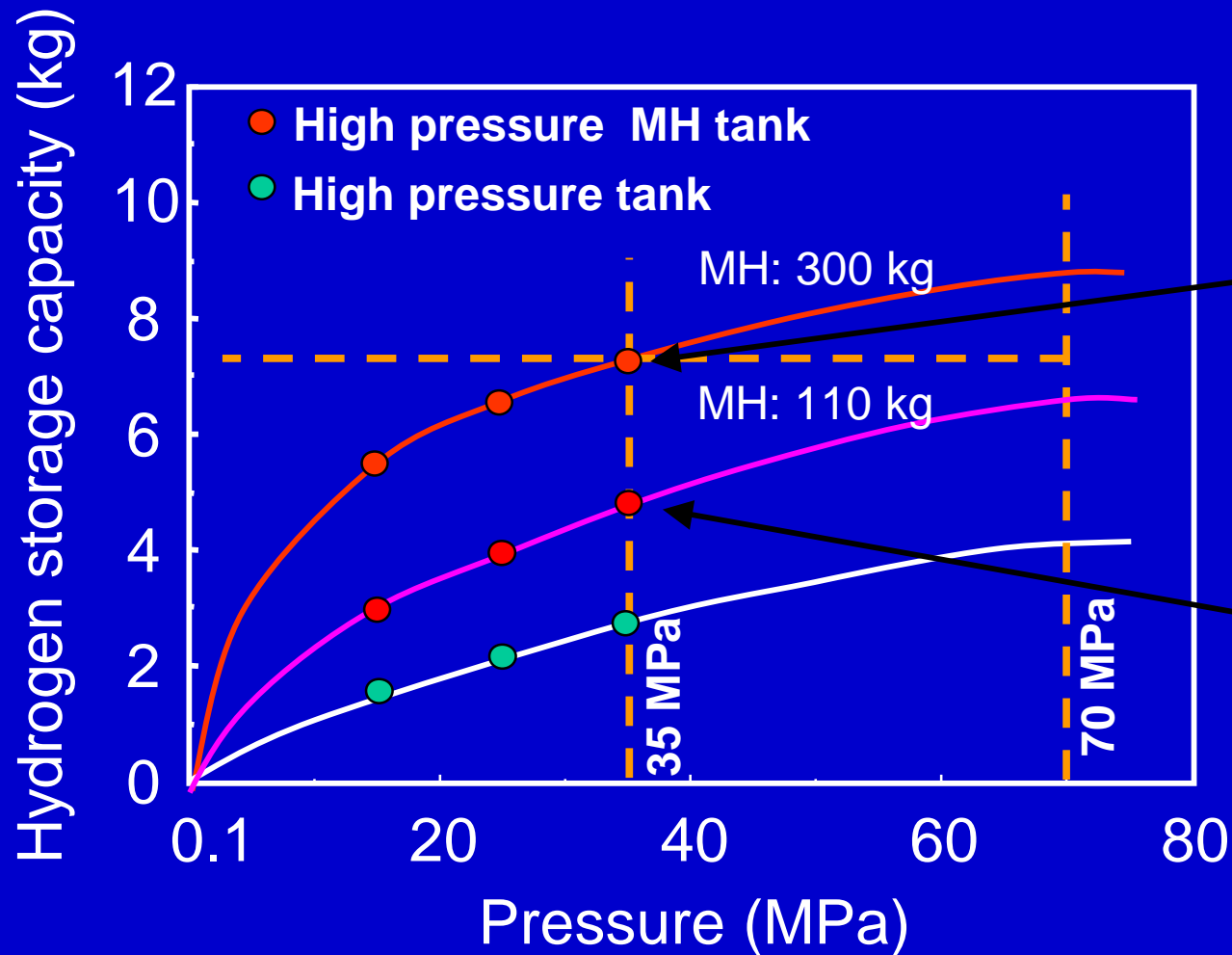


# Evaluation Method of Vehicle Scale Tank



•Charge and discharge is mainly controlled by pressure

# Results: Hydrogen Storage Capacity

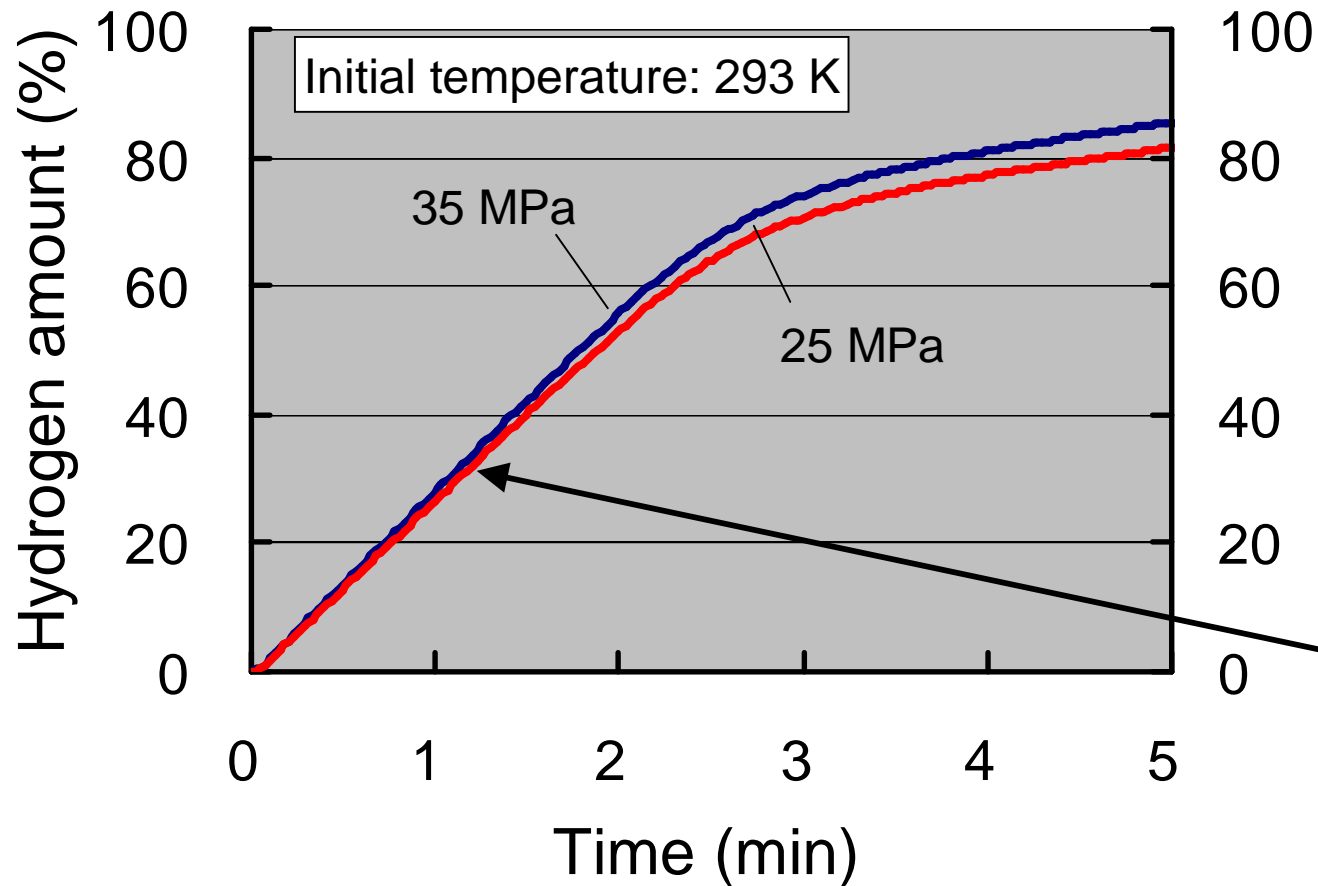


Temperature: 293 K  
Tank volume: 180 L

$H_2/MH = 2.4 \text{ mass\%}$

$H_2/MH = 4.5 \text{ mass\%}$

# High Speed Charge of Hydrogen

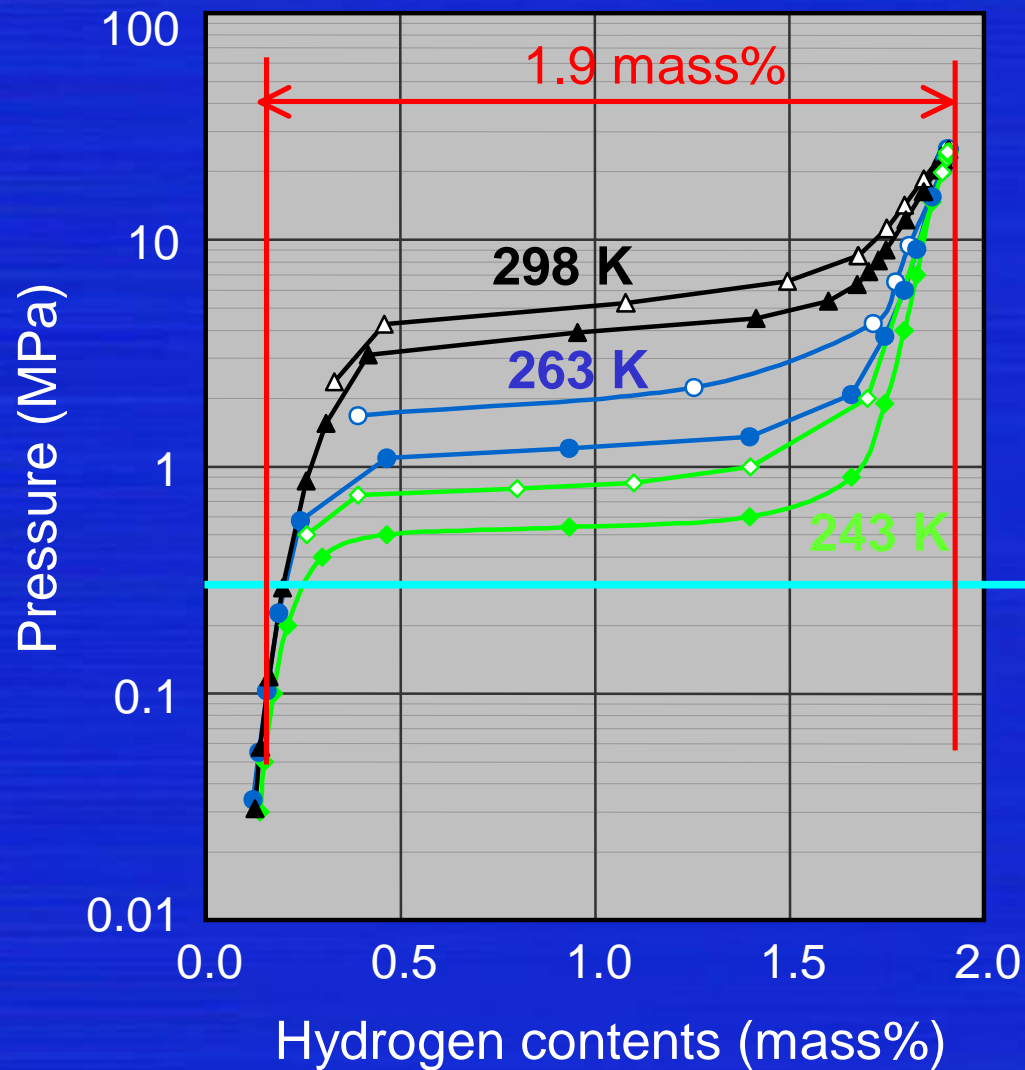


Tank number: 4  
Volume: 180 L  
Hydrogen: 5 kg  
(at 35 MPa)

H<sub>2</sub> flow rate:  
<11000 NL/min



# Desorbing Ability at Low Temperature



Absorption (298 K)  
Desorption (298 K)  
Absorption (263 K)  
Desorption (263 K)  
Absorption (243 K)  
Desorption (243 K)

Required pressure to supply H<sub>2</sub> to FC system

Fig. PCT diagram of Ti-Cr-Mn alloy

# Performance of On-board Tank System

	Low-pressure MH tank Ti-Cr-V System	High-pressure tank	High-pressure MH tank Ti-Cr-Mn System
Hydrogen storage capacity	3.5 kg / tank 120 L	3 kg / tank 180 L	7.3 kg / tank 180 L
Tank weight	300 kg	< 100 kg	420 kg
Hydrogen filling time	30-60 min. With external cooling facility	5-10 min.	5 min. / 80 % Equal to high-pressure tank without cooling facility
Hydrogen release at low temperature	Difficult under 308 K	Possible	Possible even at 243K
Control ability	Difficult in acceleration	Good	Good Equal to high-pressure tank
Safety	Low pressure (< 1 MPa)	High-pressure (35 MPa)	High-pressure (35 MPa)

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# Target Performance for Metal Hydrides

Item	Specification	Note
1. Hydrogen storage density	Weight > 3-4 mass% Volume (V/V <sub>0</sub> ) > 1,800-2,400	V = stored hydrogen gas volume (273K, 1atm) V <sub>0</sub> = volume of MH
2. Enthalpy	ΔH   < 20 kJ/molH <sub>2</sub>	
3. Equilibrium pressure	> 1.0 MPa / 243 K (desorbing) < 35 MPa / 393 K (absorbing)	
4. Cyclic durability	Decrease of storage capacity < 10% / 1,000 cycles < 5% / 100 cycles	H <sub>2</sub> purity > 99.99 %

# Recent Activities about Hydrogen Storage

- 1) D. Mori, N. Haraikawa, N. Kobayashi, T. Shinozawa, T. Matsunaga, H. Kubo, K. Toh and M. Tsuzuki, "High-pressure Metal Hydride Tank for Fuel Cell Vehicles", 2005 MRS Spring Meeting
- 2) D. Mori, N. Kobayashi, T. Shinozawa, T. Matsunaga, H. Kubo, K. Toh and M. Tsuzuki, J. Japan Inst. Metals, 69, 308 (2005)
- 3) D. Mori, N. Kobayashi, T. Matsunaga, K. Toh and Y. Kojima, Materia Japan, 44, 257 (2005).
- 4) T. Matsunaga, T. Shinozawa, H. Suzuki and D. Mori, "High Desorption Pressure Metal Hydride for High-pressure MH Tank", E-MRS 2005 SPRING MEETING
- 5) H. Suzuki, T. Mouri, K. Tange, Y. Kojima, "Development of Hydrogen Storage Materials for Fuel Cell Vehicle#", ICMAT & ICAM 2005, 3-8 July 2005, Singapore, SYPOSIA (P) Materials for Rechargeable Batteries, Hydrogen Storage and Fuel Cell



## *Summary*

### **-Performance of High-pressure MH System**

#### **1. Hydrogen storage capacity**

**max.7.3kg / tank (volume 180L)**

#### **2. High speed charge**

**hydrogen charging rate is over 11,000NL/min**

**(same as 35MPa cylinder vessel)**

#### **3. Release H<sub>2</sub> at low temperature from 243K**

**-High-pressure MH system shows a realistic way to obtain adequate cruising range over 700km.**

**-Large gap to target performance is still remained.**

**To realize hydrogen society, worldwide collaboration study is expected in this field.**



**Sustainable Mobility**  
**TODAY for TOMORROW**

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