

Materials for High-Temperature Thermochemical Processes

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OAK RIDGE NATIONAL LABORATORY
U. S. DEPARTMENT OF ENERGY

NHI Materials for TC Processes Task Focuses on Planning & Coordinating

Timeline

- ❖ Project start date FY04
- ❖ Project end date FY08
- ❖ Percent complete ~10%

Budget

- ❖ Total project funding – 1M\$
 - ◆ DOE share: 1M
 - ◆ Funding received in FY04: 0.05M
- ❖ FY05 Funding: 0.1M

Barriers

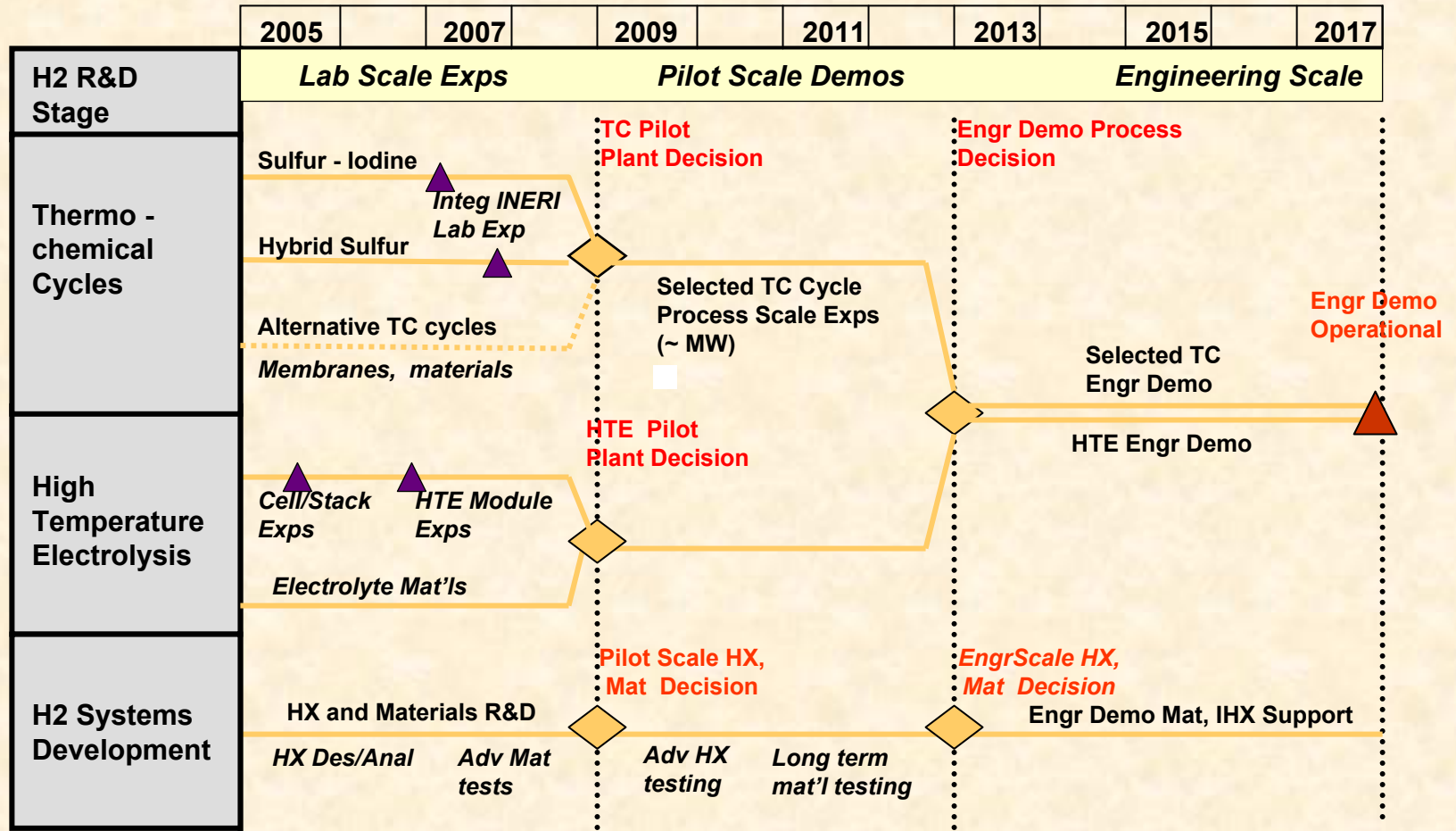
- ❖ Thermochemical (TC) process service environments
 - ◆ High temperature
 - ◆ Corrosive
 - ◆ Lack of sufficient testing capabilities for TC conditions

Partners

NHI TC Program

- ❖ Industry (GA, Cerametec)
- ❖ Universities (UNLV, MIT, UCB)
- ❖ Labs (INL, ORNL, SNL)

NHI Materials for TC Processes Plans and Coordinates R&D through FY 2008



Objectives of Materials for High-Temperature Thermochemical Processes Are:

- ❖ Assess range of service conditions for NHI Thermochemical processes (Sulfur cycles, Ca-Br, others)
- ❖ Identify candidate materials of construction for cycle components (alloys, ceramics, refractories)
- ❖ Develop materials testing approach and priorities to support NHI TC cycle development (NHI Materials Development and Testing Plan)
- ❖ Coordinate materials planning for NHI and monitor “evolving” research and development activities

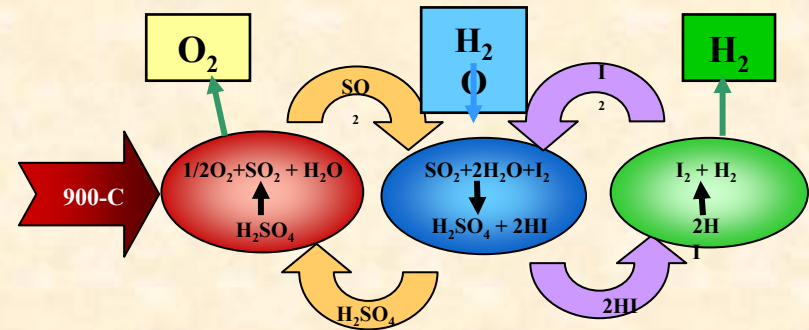
System Interface and High Temperature Heat Exchanger Programs at INL & UNLV are part of this activity (S. Sherman, INL, coordinates)

Thermochemical Cycle R&D Areas Include:

❖ Thermochemical Cycles (Scaling, efficiency)

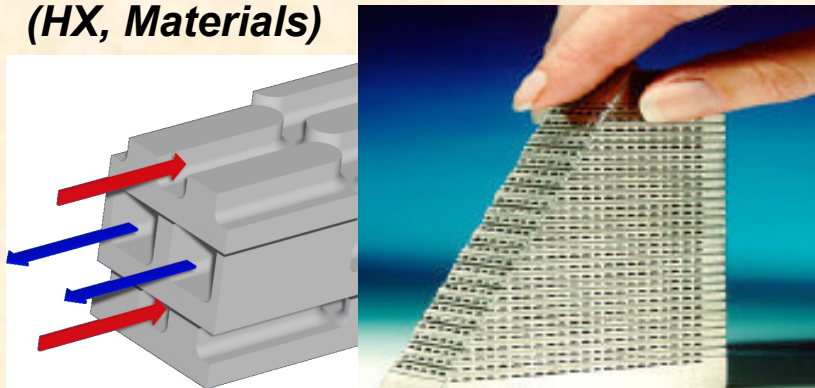
- ◆ Sulfur cycles (S-I, Hybrid)
- ◆ Alternate, Ca-Br

❖ System Interface (High temp materials and HX design)

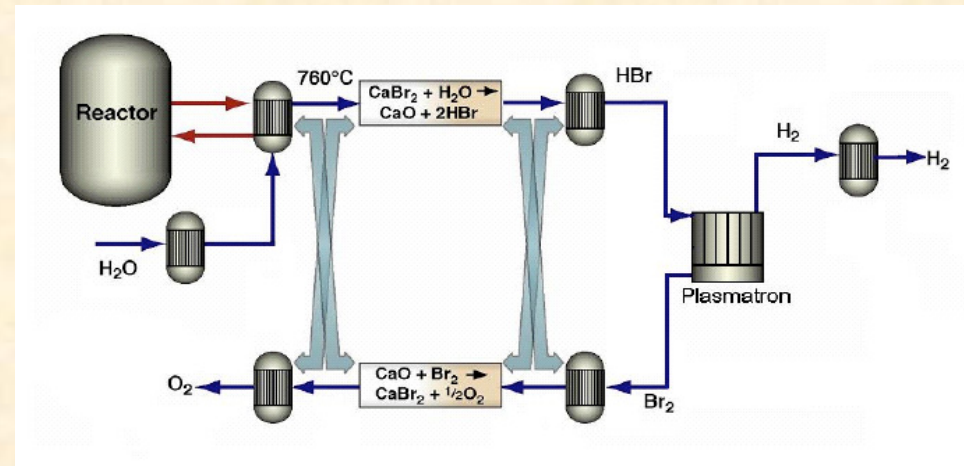


S-I Thermochemical Cycle

Interface Technologies (HX, Materials)



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Calcium Bromine Cycle

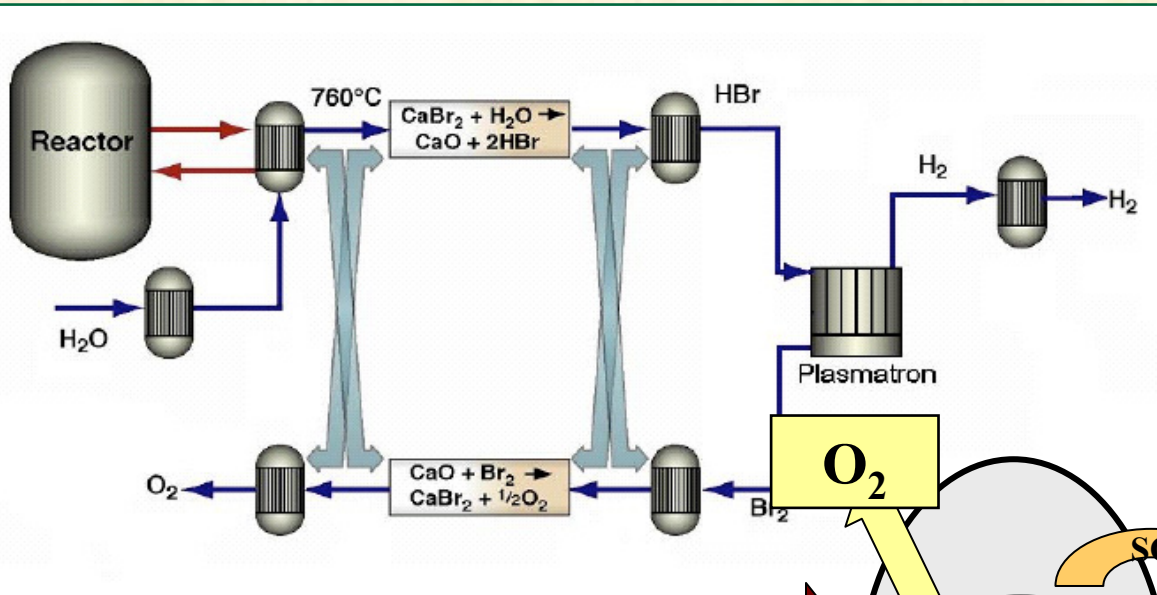
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Other TC Cycles with Potential for Lower Temperature or Higher Efficiency Include:

- ❖ **Copper chlorine (GRI H-6)**
 - ◆ <550°C; low peak temperature
 - ◆ Issue: higher efficiency electrolysis
- ❖ **Iron chloride (Ispra Mark 9, GRI I-6)**
 - ◆ 650°C; low peak temperature
 - ◆ Issue: suppress competing chemical reaction
- ❖ **Iron chloride (GRI B-1)**
 - ◆ 925°C; more mature
 - ◆ Issue: suppress competing chemical reaction
- ❖ **Copper sulfur (GRI H-5)**
 - ◆ 827-900°C; potential for higher efficiency
 - ◆ Issue: economics of scaling hybrid processes; higher efficiency electrolysis
- ❖ **Vanadium chlorine**
 - ◆ 925°C; full flowsheet available
 - ◆ Issue: high peak temperature; conflicting data on one reaction

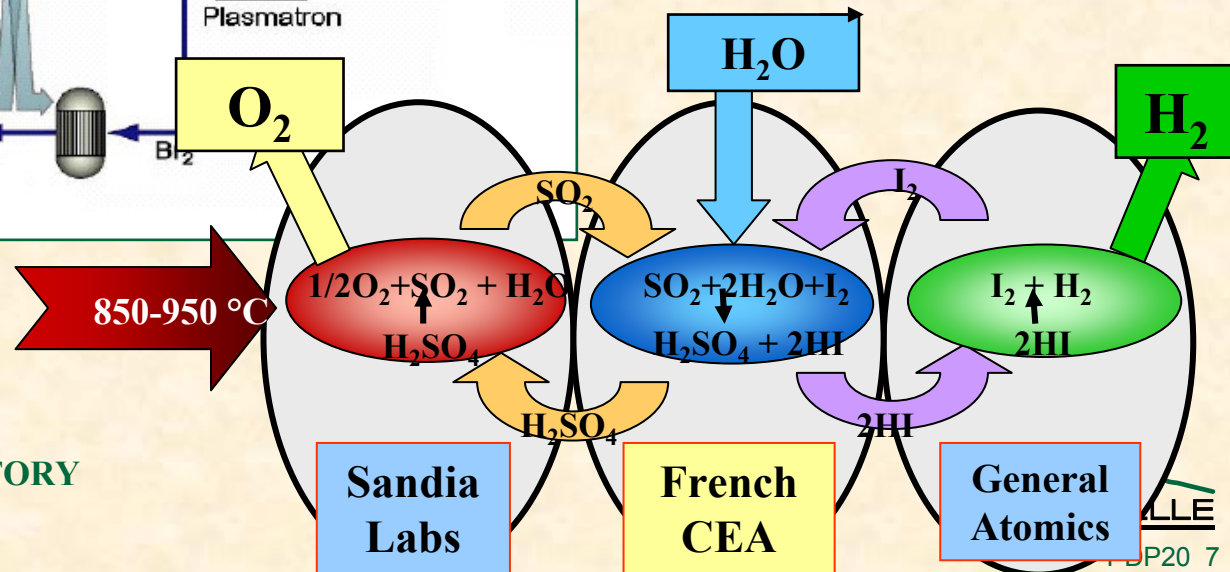
Approach to Planning & Coordinating Includes:

- ❖ Understand the chemistry and temperature of various process steps
- ❖ Identify possible materials for use in these steps
- ❖ Establish test program to evaluate the materials/process



Sulfur Iodine Cycle

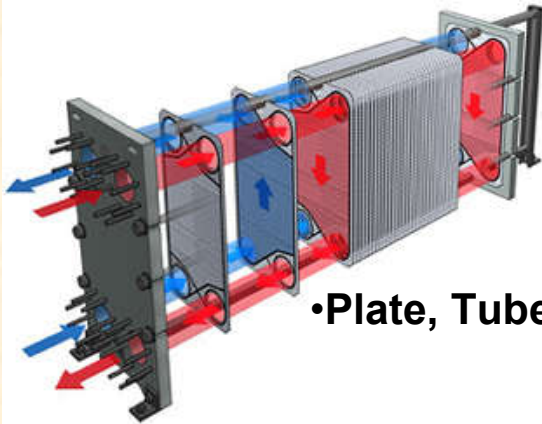
- ❖ HI section (GA)
- ❖ H₂SO₄ Section (SNL)
- ❖ Bunsen (CEA)
- ❖ Materials (INL, ORNL, Univ, Ind)



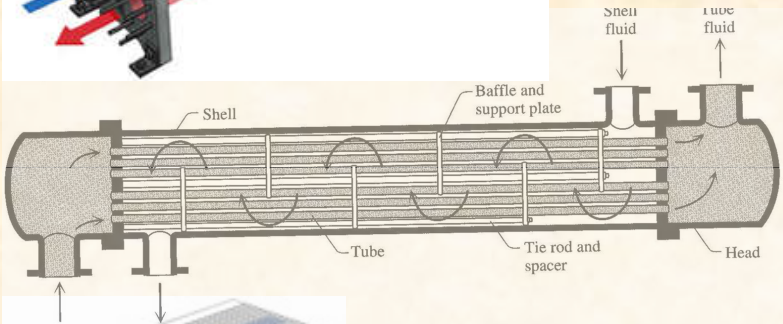
Materials for High-Temperature Thermochemical Processes - Technical Accomplishments Include:

- ❖ **Completed assessment of NHI TC cycle service conditions, candidate materials**
- ❖ **Issued report: Materials Requirements for Nuclear Hydrogen Generation Systems**
 - ◆ **Identifies chemistry and temperatures for key process steps**
 - ◆ **Identifies candidate materials for major component in NHI TC cycles**
 - ◆ **Identifies testing approach and prioritizes materials R&D needs**

Heat Exchanger Materials and Design Are Key Issues for Thermochemical Cycles

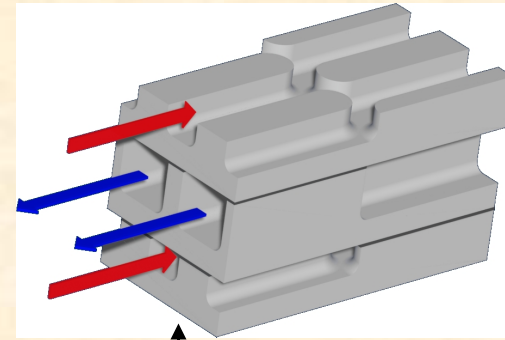


•Plate, Tube/Shell

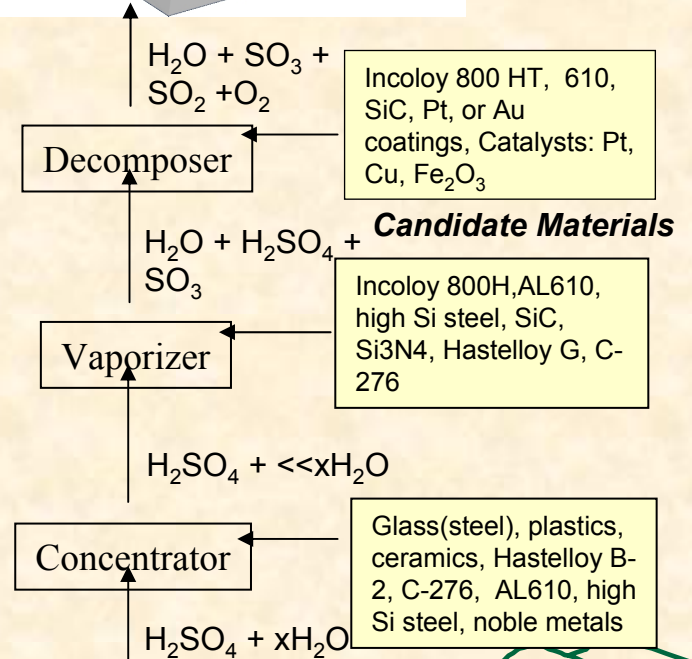


•Heatrix

•Advanced Ceramic Materials (SiC, C-SiC)



Offset-fin C-SiC Composite plate heat exchanger



Technical Accomplishment-Isolate Chemical and Temperature Environments (H₂SO₄ & HI) Generation

S-I Bunsen Reaction (120 – 130 °C; 0.1 to 0.3 MPa)

Stage	Environment (wt%)	Candidate material
Main Reaction	8 H ₂ O; 2 H ₂ SO ₄ ; 80 I ₂ ; 7 HI; 0.5 O ₂ ; 1.7 SO ₂	Ta, Zr, Si ₃ N ₄ , SiO ₂ , Al ₂ O ₃ , B2, 242, Hastelloy C-276, and Nb-1Zr coating
H ₂ SO ₄ Boost Reactor	57 H ₂ O; 43 H ₂ SO ₄ ; 0.1 SO ₂ ; trace I ₂	
HI Phase SO ₂ Stripper HX	trace H ₂ SO ₄ ; 87 I ₂ ; 7 HI; 0.2 SO ₂ ; 6 H ₂ O;	
Vessel for Bunsen Reaction HXs	Ranges listed above	Fluorocarbon-lined (Teflon, Kynar, etc.) low alloy steels

Technical Accomplishment-Isolate Chemical and Temperature Environments (H_2SO_4 Concentration)

S-I H_2SO_4 Concentrator

Stage	Temperature (°C)	Pressure (MPa)	Environment (wt%)	Candidate materials
Primary side	<450	0.1-6.8	He/molten salt	Hastelloy B2 & N, SiC, C-C composites, C276, 800/800H, Hi-Si steel (Duriclor 51M), glass-lined steel, Au, Pt, & Nb coatings
Secondary side	<400	0.1	0-0.1 SO_2 ; 57-98 H_2SO_4 ; 2-42 H_2O ; trace I_2	Hastelloy B2 & N, SiC, C-C composites, C276, 800/800H, Hi-Si steel (Duriclor 51M), glass-lined steel, Au, Pt, & Nb coatings

Technical Accomplishment-Isolate Chemical and Temperature Environments (H_2SO_4 Vaporizer)

S-I H_2SO_4 Vaporizer

Stage	Temperature (°C)	Pressure (MPa)	Environment (wt%)	Candidate materials
Primary side	580 to 380	0.1-6.8	He/molten salt	Hastelloy B2, G & N, SiC, C-C composites, Si ₃ N ₄ , C276,
Secondary side	330 to 530	0.7	Liquid-Vapor 98-71 H_2SO_4 ; 0-22 SO_3 ; 2-7 H_2O ;	800/800H, Hi-Si steel (Duriclor 51M), glass-lined steel, Au-, Pt-, & Nb-coatings

Technical Accomplishment-Isolate Chemical and Temperature Environments (H_2SO_4 Decomposer)

S-I H_2SO_4 Decomposer

Stage	Temperature (°C)	Pressure (MPa)	Environment (wt%)	Candidate materials
Primary side	950 to 800	0.1-6.8	He/molten salt	Hastelloy B2, SiC, C-C composites, Si ₃ N ₄ , 242, 214, C276, 800HT, Nb-1Zr, Au-, Pt-, Cu-, & glass-coatings, Pt-, Cu-, & Fe ₂ O ₃ -catalysts
Secondary side	530 to 900	0.7	Inlet-Outlet 71-20 H_2SO_4 ; 22-13 SO_3 ; 7-16 H_2O ; 0-40 SO_2 ; 0-10 O_2	

Technical Accomplishment-Isolate Chemical and Temperature Environments (HI Decomposition)

S-I Hlx Reactive Distillation

Stage	Temperature (°C)	Pressure (MPa)	Environment (wt%)	Candidate materials
Inlet feed stream	262	2.2	11 HI; 81 I ₂ ; 8 H ₂ O	
Outlet column bottom	310	2.2	1 HI; 98 I ₂ ; 1 H ₂ O	Ta, Ta-10W, Mo, Nb, Nb-1Zr, Zircaloy 702, SiC, vitreous carbon, C-C composites, Bulk metallic glasses
H ₂ scrubber/ condenser	221 to 25	2.2	66 HI; 32 H ₂ O; 2 H ₂	

Technical Accomplishment-Isolate Chemical and Temperature Environments (HI Decomposition)

S-I Hlx Phosphoric Acid Reactor

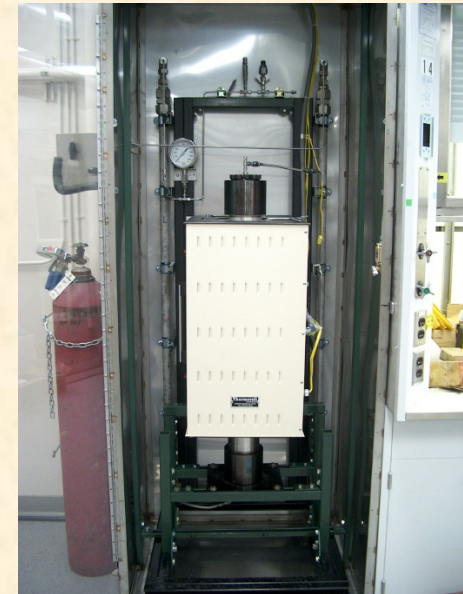
Stage	Temperature (°C)	Pressure (MPa)	Environment (wt%)	Candidate materials
Concentrated H ₃ PO ₄	132 to 211	0.1	96 H ₃ PO ₄ ; 4 H ₂ O	
Custom column	120 to 241	0.3-0.9	74 H ₃ PO ₄ ; 11 HI; 4 I ₂ ; 10 H ₂ O	TBD based on relevant industrial experience
Dilute H ₃ PO ₄	250	0.95	87 H ₃ PO ₄ ; 13 H ₂ O	
Iodine outlet	120	0.2-0.7	99.9 I ₂ ; 0.1 H ₂ O	

Materials Testing for NHI Thermochemical Cycles Has Been Initiated - HI Decomposition



❖ *Work being performed by GA*

- ◆ S-I corrosion test facilities constructed for corrosion tests on selected coupons and stress specimens
- ◆ Screening tests underway for HI section materials
- ◆ Longer term testing, advanced materials activities being defined



Immersion corrosion test system II

Materials Testing for H_x Section Has Begun

❖ Refractory metals and ceramics have shown the best corrosion performance to date

❖ H_x materials testing (UNLV and GA)

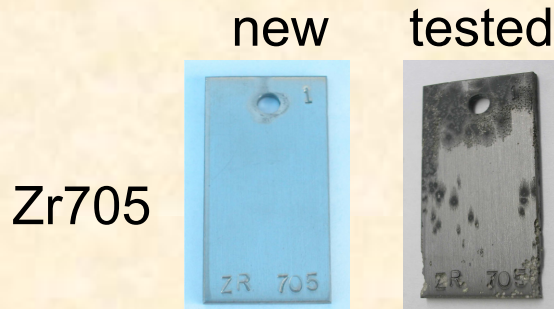
❖ 22 coupons from four classes of materials: refractory and reactive metals, superalloys and ceramics, have been screened.

Excellent	Good	Fair	Poor
Ta-40Nb, Nb-1Zr, Nb-10Hf, SiC(CVD), SiC(Ceram atec sintered), Mullite	Ta, Ta- 10W, Nb, Nb-7.5Ta, SiC (sintered) Si-SiC (3 kinds)	Mo-47Re, Alumina	Mo, C-276, Haynes 188, graphite*, Zr702, Zr705

* structurally sound but absorbed H_x

❖ Long term corrosion performance testing has started

- effect of H_x on stress corrosion
- cost reduction through cladding



Technical Accomplishment-Isolate Chemical and Temperature Environments (Ca-Br Cycle)

Reaction Beds During HBr Production (0.1 MPa)

Stage	Temperature (°C)	Environment (wt%)	Candidate materials
HX primary side	750	He/molten salt	Hastelloy N, 232, 214, 713 cast, RA330
HX secondary side		100 CaBr ₂ & H ₂ O to 100 CaO & HBr	
Bed support		Ceramics/catalysts TBD	
Reaction vessel	300 to 750	100 H ₂ O & 0 HBr to 0 H ₂ O & 100 HBr with trace O ₂	If insulated, Ni-Clad low alloy steel, If not, 713LC, 214, Ni ₃ Al, MA956, MA754
Reaction vessel insulation	750		CaTiO ₃ , Al ₂ O ₃

Technical Accomplishment-Isolate Chemical and Temperature Environments (Ca-Br Cycle)

Reaction Beds During Regeneration (0.1 MPa)

Stage	Temperature (°C)	Environment (wt%)	Candidate materials
HX primary side	<590	He/molten salt	Hastelloy N, 232, 214, 713 cast, RA330
HX secondary side		100 Br ₂ & CaO to 100 CaBr ₂ & O ₂	
Bed support		Ceramics/catalysts TBD	
Reaction vessel	200 to 590	100 Br ₂ & 0 O ₂ to 0 Br ₂ & 100 O ₂	If insulated, Ni-Clad low alloy steel, If not, 713LC, 214, Ni ₃ Al, MA956, MA754
Reaction vessel insulation	<590		CaTiO ₃ , Al ₂ O ₃

Technical Accomplishment-Isolate Chemical and Temperature Environments (Ca-Br Cycle)

Reaction Beds to Plasmatron HX (0.1 MPa)

Stage	Temperature (°C)	Environment (wt%)	Candidate materials
Primary side	750 to 50	100 HBr & 0.1 H ₂ O to 3 Br ₂ & 97 HBr with trace H ₂ O	Ni, B2, 214, 232, Hastelloy N, stainless steel, Si ₃ N ₄ , Nb-1Zr
Secondary side	25	H ₂ O	

Plasmatron HX (0.1 MPa)

Stage	Temperature (°C)	Environment (wt%)	Candidate materials
Inlet	50	3 Br ₂ & 97 HBr with trace H ₂ O	300 series stainless steel
Outlet	<<300	50 HBr, 25 Br ₂ , & 25 H ₂	

Technical Accomplishment-Isolate Chemical and Temperature Environments (Ca-Br Cycle)

Other Components

Stage	Temperature (°C)	Pressure (MPa)	Environment (wt%)	Candidate materials
Multistage compressors	100 to 335	0.005 to 3.5	30 H ₂ , 4 Br ₂ , & 2 H ₂ O	TBD based on industrial experience
Steam superheater Primary side	750 to 850	6.8-0.1	He/molten salt	617, 625, 230, B2, 214, 242, Hastelloy N
Steam superheater Secondary side	750	0.1	H ₂ O	

Summary of High Priority R&D

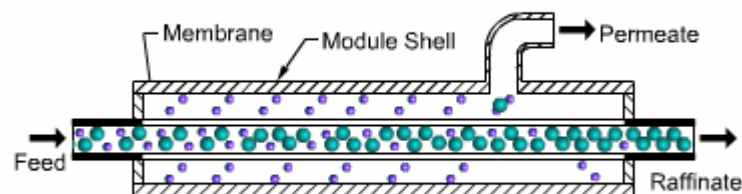
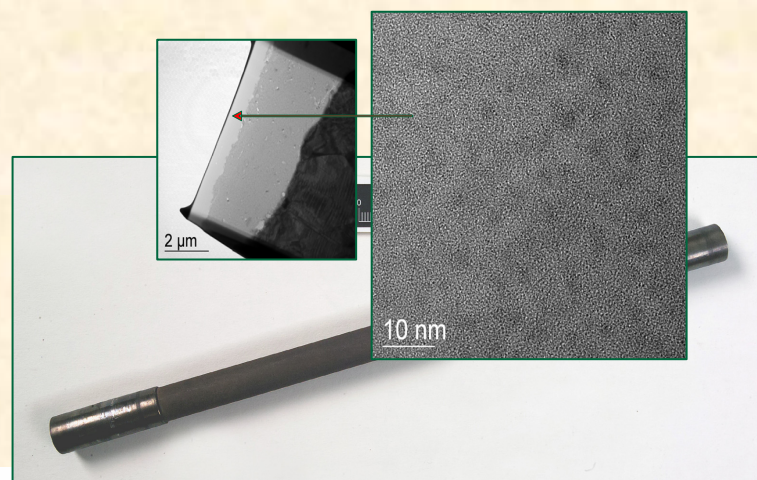
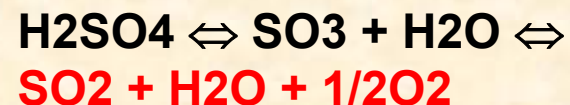
❖ Prioritized high due to technical viability issues associated with high temperature/corrosive environments

◆ Sulfur Cycles (S-I, Hybrid S)

- H₂SO₄ concentrator, vaporizer
- HI reactive distillation column
- Inorganic membranes
 - Potential for increased product formation
 - Potential for reduced operating temperature

◆ CaBr

- Reactive bed HX
- Ca-Br/HBr HX
- Reaction bed vessel



Future Work

- ❖ **FY05 – Revise and update materials selection document**

- ❖ **FY06 - Develop a prioritized, integrated materials evaluation program**
 - ◆ **So as to establish engineering feasibility**

Publications and Presentations

- ❖ **Materials Requirements for Nuclear Hydrogen Generation Systems, ORNL TM, 2004**
- ❖ **W. R. Corwin, *NHI Materials Selection and R&D Plan*, NHI Materials Planning Meeting, UNLV, Las Vegas, Nevada, May 17, 2004**
- ❖ **W. R. Corwin, *NHI Materials Selection and R&D Plan & NHI Membrane Studies Update*, NHI Semiannual Program Review, DOE HQ, Germantown, Maryland, September 21-22, 2004**
- ❖ **W. R. Corwin, *High Priority Materials R&D for NHI*, NHI Materials Program Review Meeting, ORNL, February 10, 2005**

Hydrogen Safety

What is the most significant hydrogen hazard associated with this project?

- ❖ Hydrogen explosiveness/flammability (4 to 75 vol.%) during performance evaluation

What are you doing to deal with this hazard?

- ❖ Work will be performed under the safety envelopes of the various testing organizations
- ❖ Use both administrative and physical controls
 - ◆ Work review/authorization
 - ◆ Volume/pressure controls
 - ◆ Sensors